Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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#### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.038 wR factor = 0.104 Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# (rac-5RS,7RS,8SR)-Spiro[7-methoxy-

carbonyl-1-aza-3-thiabicyclo[3.3.0]octane-8,1'-acenaphthylen]-2'-one

The title compound,  $C_{19}H_{17}NO_3S$ , crystallizes in the monoclinic space group  $P2_1/c$ , with four molecules in the unit cell. The structure contains a spiro-carbon connecting a bicyclooctane group, consisting of two fused five-membered rings, to an acenaphthylene moiety. The five-membered ring containing the S atom is in an envelope conformation. The conformation of the other five-membered ring of the bicyclo-octane moiety is close to a half-chair. The methoxycarbonyl group is equatorially substituted.

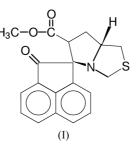
## Comment

We are interested in the X-ray structure determination of some novel polycyclic spiro-heterocyclic compounds, as they can be used as precursors for the total synthesis of naturally occurring alkaloids (Grundon, 1988). The bicyclo-octane ring system has three chiral atoms, *viz*. C5, C7 and C8 (Fig. 1), and the coordinates of these atoms correspond to 5*R*, 7*R* and 8*S*. However, the compound crystallizes as a racemic mixture in the space group  $P2_1/c$ . The angles around the spiro atom C8 are C7–C8–N1 = 103.34 (12)°, C9–C8–C18 = 102.40 (12)°, N1–C8–C18 = 116.20 (13)°, C7–C8–C9 = 111.24 (13)°, C18–C8–C7 = 116.12 (12)° and N1–C8–C9 = 107.46 (12)°. The methoxycarbonyl group is equatorially substituted at C7. The acenaphthylene moiety occupies two substituent positions at C8.

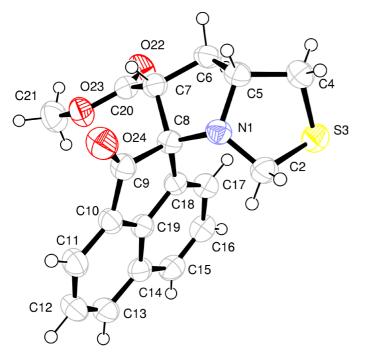
The  $Csp^2$  to  $Csp^2$  distances in the acenaphthylene group range from 1.366 (3) (C15–C16) to 1.570 (2) Å (C8–C9) and the C–C–C bond angles range from 102.40 (12) (C9–C8– C18) to 123.78 (15)° (C14–C19–C18). These values compare well with those of previously reported related structures (Hazell, 1976; Hazell & Hazell, 1977; Hazell & Weigelt, 1976; Jones *et al.* 1992).

Both the fused five-membered rings of the bicyclo-octane moiety are puckered. The N1/C2/S3/C4/C5 five-membered ring of the bicyclo-octane group is in an envelope conformation [puckering parameters Q = 0.498 (2) Å and  $\varphi = 176.2$  (2)° for the atom sequence S3-C2-N1-C5-C4; Cremer & Pople, 1975], with atom S3 in the flap position. The torsion

Received 6 November 2002 Accepted 14 November 2002 Online 22 November 2002



# organic papers



#### Figure 1

The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

angle C2-S3-C4-C5 is -39.58 (12)°, while the angle C4-S3-C2-N1 is  $38.80 (12)^\circ$ . The conformation of the other five-membered ring (N1/C5/C6/C7/C8) of the bicyclo-octane group is close to a half-chair [puckering parameters Q =0.384 (2) Å and  $\varphi = 86.4$  (3)° for the atom sequence N1–C5– C6-C7-C8]. The dihedral angle between the least-squares planes of the above two rings is  $51.96 (10)^{\circ}$ .

The C5–N1 bond distance is 1.488 (2) Å. This is comparable to the  $Csp^3$ -Nsp<sup>3</sup> distances found in similar structures (Sussman & Wodak, 1973; Wodak, 1975). The geometry around the N atom is pyramidal. The bond distance C2-S3 is 1.8323 (18) Å and S3-C4 is 1.8044 (18) Å. The angle C2-S3-C4 is 86.99 (8)°. The structure is stabilized mainly by van der Waals interactions.

# Experimental

A mixture of acenaphthylene-1,2-dione (0.364 g, 2 mmol), (R)-(-)thiazolidine-4-carboxylic acid (0.27 g, 2 mmol) and methyl acrylate (0.172 g, 2 mmol), in an equimolar ratio, was refluxed for 22 h in acetonitrile under a nitrogen atmosphere. After completion of the reaction, monitored by the disappearance of the acenaphthylene-1,2dione spot in TLC analysis, the reaction mixture was filtered to remove the unreacted acid. The filtrate was concentrated in vacuo and kept at 273 K. A yellow crystalline solid was obtained and recrystallized from chloroform/n-hexane as yellow cuboidal crystals (0.604 g, 75%; m.p. 445 K). A small amount of the compound (0.2 g) was dissolved in chloroform (2 ml). N-Hexane (8 ml) was layered carefully down the side of the test tube on to the solution. The volume ratio of solvent to precipitant was 1:4. The test tube was plugged with cotton wool and left to stand undisturbed for 24 h. As the precipitant (n-hexane) diffused into the solution, crystals were formed at the interface.

# Crystal data

C <sub>19</sub> H <sub>17</sub> NO <sub>3</sub> S	$D_x = 1.398 \text{ Mg m}^{-3}$		
$M_r = 339.40$	Cu $K\alpha$ radiation		
Monoclinic, $P2_1/c$	Cell parameters from 9012		
a = 9.3597 (1)  Å	reflections		
b = 14.7896 (2) Å	$\theta = 2.2-68.3^{\circ}$		
c = 14.9199 (1) Å	$\mu = 1.93 \text{ mm}^{-1}$		
$\beta = 128.660 \ (5)^{\circ}$	T = 293 (2) K		
V = 1612.73 (3) Å <sup>3</sup>	Rectangular block, yellow		
Z = 4	$0.32 \times 0.20 \times 0.18 \text{ mm}$		

2567 reflections with  $I > 2\sigma(I)$ 

 $R_{\rm int} = 0.051$  $\theta_{\rm max} = 68.2^{\circ}$ 

 $h=-11\rightarrow 11$ 

 $k = -17 \rightarrow 17$ 

 $l = -16 \rightarrow 17$ 

# Data collection

Nonius KappaCCD diffractometer  $\varphi$  scans with  $\kappa$  offsets Absorption correction: multi-scan (SORTAV; Blessing, 1995)  $T_{\min} = 0.583, T_{\max} = 0.707$ 21 038 measured reflections 2926 independent reflections

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0531P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.038$	+ 0.4649P]
$wR(F^2) = 0.104$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} < 0.001$
2926 reflections	$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
218 parameters	$\Delta \rho_{\rm min} = -0.34 \mathrm{e} \mathrm{\AA}^{-3}$
H-atom parameters constrained	

# Table 1

Selected geometric parameters (Å, °).

C6-C7	1.517 (2)		
C2-N1-C8	119.47 (13)	C8-N1-C5	110.46 (12)
C2-N1-C5	111.25 (13)	C7-C6-C5	102.89 (13)
N1-C5-C6-C7	33.55 (16)	C5-C6-C7-C20	-163.65 (14)
C4-C5-C6-C7	152.37 (14)	C6-C7-C8-N1	30.13 (15)

All H atoms were treated as riding; C-H = 0.93-0.98 Å and  $U_{iso}(H) = 1.5U_{eq}(C)$  for methyl H and  $U_{iso}(H) = 1.2U_{eq}(C)$  for all others.

Data collection: COLLECT (Nonius, 1997-2000); cell refinement: HKL SCALEPACK (Otwinowski & Minor, 1997); data reduction: HKL DENZO (Otwinowski & Minor, 1997) and SCALEPACK; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: WinGX (Version 1.64.02; Farrugia, 1999) and PLATON (Spek, 2002); software used to prepare material for publication: SHELXL97.

TVS thanks the UGC, India, for a grant in the form of Minor Research Project (F.MRP-039/02, Link No. 1030). SGG thanks the Spanish MCyT (BQU2000-0219) and FICYT(PR-01-GE-4) for financial support.

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